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## Study on Carbon and Oxygen Isotope Composition of the Calcites in Chahe Copper Deposit, Yuanjiang, Yunnan

WANG Ye, XU Zhengqi\*, LI Ping, SONG Hao

*Chengdu University of Technology, Chengdu, 610059, China*

Chahe copper deposit, located at the Yuanjiang county, Yunnan province, is a copper dominated polymetallic mineral deposit. Calcite is one of the gangue minerals and its formation is closely with the metallic mineralization. The composition of carbon and oxygen isotope in the calcites is a very useful method to trace the source of relevant fluids (Zheng et al., 2000; Wu, 2010). This paper aims to trace the source of mineralization fluid by the study on carbon and oxygen isotope composition of the calcites in copper deposit.

### 1 Geology of Deposit

Chahe copper deposit is situated in Dahongshan dome of the depression of the central Yunnan province, which is surrounded by Kandian axis of earth, west of Yangtze paraplatform, northeast of Honghe deep-fault and Lvzhijiang deep-fault striking south-north. The emergence strata is a suit of mid-deep metamorphic volcanic sedimentary, whose age is lower proterozoic. The ore body is hosted in the second lithological section, which is shaped as layer, like-layer, lentoid. And the ore body is within the quartzite, mica-quartz schist. The main ore minerals are chalcopyrite, pyrite, magnetite, pitchblende. The gangue minerals are quartz and calcites.

Silicification is prevalent and strong in the country rocks, there also have chloritization, carbonatation, micacization. Silicification occurs in the whole process of hydrothermal mineralization and the degree of Silicification is stronger than carbonatation in the wall rocks.

### 2 Composition of Carbon and Oxygen Isotope in the Calcites

Calcite is one of the gangue minerals in Chahe copper deposit, which occurs as veins, massed, sheet. According to the analytical results of trace elements, we divide the

samples into calcites from two different sources. One, like sample C02-2, have the same origin with the country rocks, which have the same standard pattern with the wall rock; the other, such as C02-1, C14 and C15, have the uniform standard pattern, which absolutely differ from the standard pattern of C02-2 and the country rocks.

The carbon and oxygen isotope compositions of the two periods of calcites in copper deposit, Chahe are listed in table 1. The calcites featured with high  $\delta^{18}\text{O}$  and low  $\delta^{13}\text{C}$ . The sample C02-1, C14 and C15 have limited variation of  $\delta^{13}\text{C}_{\text{V-PDB}}$  and  $\delta^{18}\text{O}_{\text{SMOW}}$  which is  $-0.96\text{\textperthousand} \sim -3.00\text{\textperthousand}$  (average  $-1.75\text{\textperthousand}$ ),  $8.00\text{\textperthousand} \sim 8.89\text{\textperthousand}$  (average  $8.56\text{\textperthousand}$ ) respectively. Sample C02-2, the same source as the country rock, have higher and lower increasing range value of  $\delta^{13}\text{C}_{\text{V-PDB}}$  and  $\delta^{18}\text{O}_{\text{SMOW}}$ , than sample C02-1, C14 and C15.  $\delta^{13}\text{C}_{\text{V-PDB}}$  and  $\delta^{18}\text{O}_{\text{SMOW}}$  of C02-2 is  $-0.84\text{\textperthousand}$  and  $8.96\text{\textperthousand}$  separately.

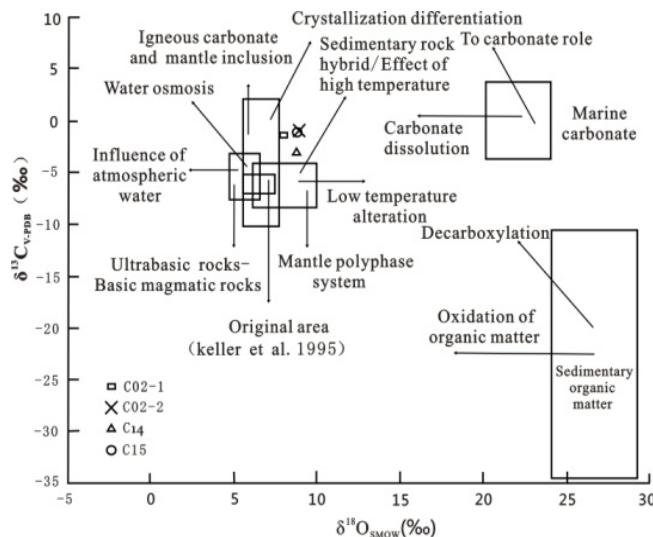
The composition of carbon isotope is an effective method to judge the source of  $\text{CO}_2$  in fluid. It is well known that the sources of carbon in mineralization fluid maybe mantle, sedimentary carbonate, organic carbon (Feng et al., 2011). The composition of carbon isotope in calcites is close to the heterogeneous system of the mantle (Figure 1), but the composition of carbon isotope of the heterogeneous system of the mantle may be affected by carbon from contamination of sedimentary rock and fractional carbon in high temperature, while the composition of oxygen is still. It is consistent with the analytical result of the composition of oxygen and carbon isotope ( $-0.84\text{\textperthousand} \sim -3.00\text{\textperthousand}$ ). This indicates that the composition of oxygen and carbon isotope are from the heterogeneous system of the mantle may be affected by carbon from contamination of sedimentary rock and fractional carbon in high temperature.

The carbon isotope composition of oceanic carbonate is  $0\pm4\text{\textperthousand}$  (Veizer, Hoefs, 1976; Hoefs, 1997). The composition of carbon isotope in calcite in Chahe is  $-$

\* Corresponding author. E-mail: 316622426@qq.com

**Table1 the composition of carbon and oxygen isotope in the calcites, Chahe Cu-deposit**

Aample	Rock style	$\delta^{13}\text{C}_{\text{V-PDB}} (\text{\textperthousand})$	$\delta^{18}\text{O}_{\text{V-PDB}} (\text{\textperthousand})$	$\delta^{18}\text{O}_{\text{SMOW}} (\text{\textperthousand})$
C02-1	calcite	-1.29	-22.22	8.00
C02-2	calcite	-0.84	-21.29	8.96
C14	calcite	-3.00	-21.47	8.78
C15	calcite	-0.96	-21.36	8.89
Average		-1.52	-21.59	8.66

Fig.1  $\delta^{13}\text{C}_{\text{V-PDB}} (\text{\textperthousand})$ - $\delta^{18}\text{O}_{\text{SMOW}} (\text{\textperthousand})$  in the Chahe Cu-deposit (modified after Zhou et al., 2012)

0.84‰-3.00‰. This is same as the carbon isotope composition of oceanic carbonate.

However, the oxygen isotope composition of oceanic carbonate is lower than it affected by dissolution (Liu., 2004). The composition of oxygen isotope in calcite in Chahe is 8.00‰~8.89‰. this imply that the source of carbon in calcite is oceanic carbonate impacted by dissolution.

### 3 Conclusion

Above all, based on the composition of carbon and oxygen of calcite in Chahe copper deposit, Yunnan, it concluded that the mineralization maybe the heterogeneous system of the mantle affected by

contamination of sedimentary rock and fractionation in high temperature, then experienced dissolution of oceanic carbonate.

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