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Estimation of Reservoir Temperature Using Silica and Cationic Solutes Geothermometers: A Case Study in the Tengchong Geothermal Areas

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Reservoir temperature estimation is vitally important for assessing the geothermal exploitation potential of the Tengchong hydrothermal area. In this study, the concentrations of major chemical constituents in geothermal waters sampled from the boiling and hot springs in Tengchong were measured, based on which quartz and cationic solutes geothermometers were used to calculate the subsurface temperatures.Whether а geothermal water sample is in full equilibrium with the reservoir minerals is crucial for the selection of suitable geothermometers. Thus, log (Q/K) diagram and Na-K-Mg triangular diagram were applied to evaluating the equilibrium status between geothermal water and reservoir hostrocks. The results show that the samples RH01, RH03, RH04, RH05 and LL16 were in or very close to full equilibrium with the selected minerals, and Na-K geothermometer is reliable for them. K/Mg geothermometer, instead, is applicable to LP08 and PZH18 whose chemical compositions were adjusted to the shallow reservoir temperatures during their re-equilibrium processes. Generally, cationic solutes geothermometers are unsuitable for the samples SQ20 and RH07 located in



Fig. 1 Simplified map of Tengchong and sampling locations

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Fig. 2 Triangular Na-K-Mg diagram for geothermal water samples from Tengchong

the immature water area of the Na-K-Mg diagram, and quartz geothermometer was adopted to evaluate their corresponding subsurface temperatures. According to the reservoir temperature estimation made in this study, there is a high-temperature reservoir with a possible temperature range of 210 to 270 $^{\circ}$ C below Rehai.

Key words: chemical equilibrium; geothermometer; geothermal springs; Tengchong; Rehai

References

- Anzil, P.A., Guereschi, A.B. and Martino, R.D., 2012. Mineral chemistry and geothermometry using relict primary minerals in the La Cocha ultramafic body: A slice of the upper mantle in the Sierra Chica of Cordoba, Sierras Pampeanas, Argentina. *Journal of South American Earth Sciences*, 40: 38-52.
- Asta, M.P. et al., 2012. Hydrochemistry and geothermometrical modeling of low-temperature Panticosa geothermal system (Spain). *Journal of Volcanology and Geothermal Research*, 235: 84-95.
- Dulanya, Z., Morales-Simfors, N. and Sivertun, A., 2010. Comparative study of the silica and cation geothermometry of the Malawi hot springs: Potential alternative energy source. *Journal of African Earth Sciences*, 57(4): 321-327.

Giggenbach, W.F., 1988. Geothermal Solute Equilibria -



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Fig. 3 Log (Q/K)-T diagrams for geothermal water samples

Derivation of Na-K-Mg-Ca Geoindicators. *Geochimica et Cosmochimica Acta*, 52(12): 2749-2765.

- Gokgoz, A. and Tarcan, G., 2006. Mineral equilibria and geothermometry of the Dalaman-Koycegiz thermal springs, southern Turkey. *Applied Geochemistry*, 21(2): 253-268.
- Guo Qinghai and Wang Yanxin, 2012. Geochemistry of hot springs in the Tengchong hydrothermal areas, Southwestern China. *Journal of Volcanology and Geothermal Research*, 215: 61-73.
- Majumdar, N., Mukherjee, A.L. and Majumdar, R.K., 2009. Mixing hydrology and chemical equilibria in Bakreswar geothermal area, Eastern India. *Journal of Volcanology and Geothermal Research*, 183(3-4): 201-212.
- Mohammadi, Z., Bagheri, R. and Jahanshahi, R., 2010.



Hydrogeochemistry and geothermometry of Changal thermal springs, Zagros region, Iran. *Geothermics*, 39(3): 242-249.

Aragonite

- Mutlu, H., 1998. Chemical geothermometry and fluid-mineral equilibria for the Omer-Gecek thermal waters, Afyon area, Turkey. *Journal of Volcanology and Geothermal Research*, 80 (3-4): 303-321.
- Pirlo, M.C., 2004. Hydrogeochemistry and geothermometry of thermal groundwaters from the Birdsville Track Ridge, Great Artesian Basin, South Australia. *Geothermics*, 33(6): 743-774.
- Reed, M. and Spycher, N., 1984. Calculation of Ph and Mineral Equilibria in Hydrothermal Waters with Application to Geothermometry and Studies of Boiling and Dilution. *Geochimica et Cosmochimica Acta*, 48(7): 1479-1492.
- Shangguan Zhiguan, Zhao Ciping, Li Hengzhong, Gao Qingwu

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No.	Location	Date	Sample type	Tem (°C)	pН	Hydrochemic al type	EC(µs/cm)
RH01	Dagunguo	2012.07.16	Hot spring	96.6	8.34	Na-HCO ₃ -Cl	4210
RH03	Huai taijing-L	2012.07.16	Hot spring	88.0	8.35	Na-HCO ₃ -Cl	3350
RH04	Huait aijing-R	2012.07.16	Hot spring	88.0	7.61	Na-HCO ₃ -Cl	2464
RH05	Gumingquan	2012.07.16	Hot spring	96.0	8.87	Na-HCO ₃ -Cl	3560
RH07	Bapai	2012.07.17	Hot spring	20.1	7.65	Na-Ca-HCO3	163.6
LP08	Langpu Dagunguo	2012.07.17	Hot spring	70.0	8.16	Na-HCO ₃ -Cl	3020
LL16	Banglazhang	2012.07.20	Hot spring	66.0	8.77	Na-HCO3	1021
PZH18	Xia ot ang	2012.07.20	Hot spring	96.0	8.00	Na-HCO ₃	1078
SQ20	Shiqiang	2012.07.21	Hot spring	62.0	6.93	Na-Ca-HCO3	2456

 Table 1
 Characteristics of water samples from Tengchong

Table 2 Concentrations of major chemical constituents in water samples from Tengchong (mg/kg)

No.	Alkalinity	SiO ₂	SO_4^{2}	Cl-	F	NO ³⁻	Ca	Mg	Na	K	Li
RH01	987.1	700	35.2	725.0	18.2	92.5	0.97	0.02	689.0	122.8	0.00
RH03	770.7	304	20.1	558.6	13.8	19.6	1.54	0.06	5383	96.8	0.00
RH04	515.2	491	38.3	454.6	8.1	15.4	1.61	0.05	400.4	71.1	0.00
RH05	826.5	483	18.6	651.0	16.1	5.6	1.38	0.03	5732	107.1	0.00
RH07	75.0	55	8.0	45.0	0.0	14.1	10.33	8.21	9.1	4.9	0.01
LP08	1217.9	150	28.2	336.9	11.7	5.1	3.84	0.98	573.8	55.9	2.89
LL16	505.2	172	38.9	54.5	19.3	0.0	1.36	0.01	203.5	15.2	3.03
PZH18	445.2	244	30.0	83.4	11.7	5.3	1.67	0.13	2099	19.0	1.44
SQ20	1623.1	79	1.8	82.8	3.3	2.6	70.45	28.08	395.5	68.9	2.03

Table 3 Expressions for different geothermometers

Geothermometer	Expression	
Quartz(a)	$\theta_{\text{SiO}_2}^{\circ} C = \frac{1309}{5.19 - \log \text{SiO}_2} - 273.15$	(1)
Quartz(b)	$\theta_{so_2}^{\circ}C = \frac{1522}{5.75 - \log SiO_2} - 273.15$	(2)
Na/K(Fournier)	$\theta_{\text{Na}/K}$ °C = $\frac{1217}{\log(Na/K) + 1.483} - 273.15$	(3)
K ² /Mg	$\theta_{K^2/Mg}^{\circ} C = \frac{4410}{14 - \log(K^2/Mg)} - 273.15$	(4)

Table 4 Reservoir temperatures calculated using different geothermometers

No.	Location	Temperature/°C					
		Quartz(a)	Quartz(b)	Na/K	K-Mg		
RH01	Dagunguo	285.08	250.79	272.10	264.42		
RH03	Huaitaijing-L	210.39	192.70	273.04	227.42		
RH04	Huai taijing-R	250.68	224.41	271.64	217.37		
RH05	Gumingquan	249.19	223.25	277.15	248.85		
RH07	Bapai	106.31	106.44	422.75	52.91		
LP08	Langpu Dagunguo	161.17	152.71	214.78	147.08		
LL16	Banglazhang	170.21	160.15	193.29	176.87		
PZH18	Xiaotang	193.97	179.52	208.63	144.03		
SQ20	Shiqiang	124.90	122.32	269.66	101.46		

and Sun Mingliang, 2005. Evolution of hydrothermal explosions at Rehai geothermal field, Tengchong volcanic region, China. *Geothermics*, 34(4): 518-526.

Verma, M.P., 2012. QrtzGeotherm: A revised algorithm for quartz solubility geothermometry to estimate geothermal reservoir temperature and vapor fraction with multivariate analytical uncertainty propagation. *Computers & Geosciences*, 48: 316-322. Zheng Xilai, Armannsson, H., Li Yongle and Qiu Hanxue, 2002. Chemical equilibria of thermal waters for the application of geothermometers from the Guanzhong basin, China. *Journal* of Volcanology and Geothermal Research, 113(1-2): 119-127.

Shangguan Zhiguan, 2000. Structure of geothermal reservoirs and the temperature of mantle-derived magma hot source in the Rehai area, Tengchong. *Acta Petrologica Sinica*, 16(1): 83-90