



The Relationship between Alteration Characteristics and Uranium Mineralization in Zhiluo Formation, North-Eastern Ordos Basin

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Abstract: The sandstone in Zhiluo Formation experienced the early oxidation at first and oxidized to brick red and purplish red sandstone when after the sediment. Then it also experienced the secondary reduction of oil and gas, changing the color of sandstone to green (Qi et al., 2007). The direct evidence was that the green sandstone wrapped the brick red and purplish red sandstone, and the residue of oxidized sandstone still could be seen in the core. The green alteration sandstone is the product of strong reducing environment forming by the secondary reduction of ancient oxidized sandstone (Wu et al., 2006). After the previous research by the author on the geochemistry characteristic of sandstone in this region, results showed that the residual oxidized purplish red and brick red sandstone had features of low content of U, S and TOC and high content of Ca and $\text{Fe}^{3+}/\text{Fe}^{2+}$; the secondary reducing green sandstone had features of low content of U, S, TOC, Ca and $\text{Fe}^{3+}/\text{Fe}^{2+}$ (Yi et al., 2015). The content of S and TOC in both two sandstones was low indicating the oxidized feature. Because the residual ancient oxidized sandstone remained the oxidation characteristic, the sandstone had the high content of $\text{Fe}^{3+}/\text{Fe}^{2+}$. The secondary reducing sandstone was influenced by the late secondary reduction of gas and oil, so the content of $\text{Fe}^{3+}/\text{Fe}^{2+}$ was low. The ore-forming process of Zhiluo Formation in this region experienced the early acidic oxidation and the late alkaline reduction process (Fan et al., 2007). In the process of the acidic oxidation, the water-rock reaction occurred between the oxygenated water in the provenance and the formation, resulting in the feldspar dissolution. The feldspar alteration released Ca^{2+} for kaolinite and activated U in the formation at the same time, altogether entering into the oxygenated water and migrating with water. As the increment of migration distance, the metal sulfides in the formation was oxidized to the sulfuric acid by the oxygen in water with the participation of aerobic bacterium, which decreased the oxygen content in water. When migrating to the redox zone, the oxygen in water ran out and the formation was in an anaerobic environment. With the participation of the sulfate-reducing bacteria, SO_4^{2-} in the formation was reduced to H_2S gas and CO_3^{2-} . CO_3^{2-} and Ca^{2+} in water reacted, formed CaCO_3 and precipitated down. Meanwhile, U in water was reduced by H_2S , leading to the precipitation and concentration

and forming the phenomenon of the concurrent precipitation of the uranium minerals and calcite. The phenomenon of the coffinite occurrence around the calcite observed under a microscope and the high content of CaO in the ore-bearing sandstone in this region both proved points mentioned above (Yi et al., 2015). Besides, the result of the C-O isotope data showed that the carbonate cement was correlated with the precipitated organic matters, and a part of CO_3^{2-} was formed by the decarboxylation of the precipitated organic matters. The basalt eruption happened at the period of the Lower Cretaceous in the research zone leads to the transformation effect of the thermal fluid in the Zhiluo Formation, forming the thermal genetic minerals and promoting the process of the uranium metallogenesis in this period. In summary, after the Zhiluo Formation in the target zone in this region precipitated, it experienced a bunch of alteration processes which were the early acidic oxidation alteration, mid-life hydrothermal fluid alteration and late secondary reduction alteration. The alteration had a wide range and high strength, and plenty alterations had a close connection with the uranium metallogenesis.

Key words: uranium mineralization, alteration characteristics, Ordos Basin

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